



THE REACTIVITY OF WOODY ASH IN COMPARISON WITH SLAG

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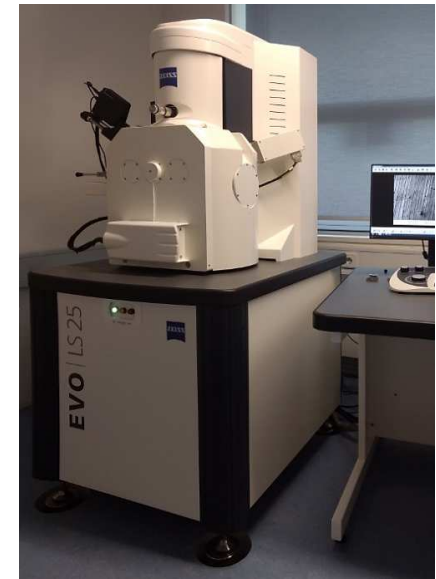
Germany

Introduction

TARECO²

Content

- 1) Introduction
- 2) Experimental part
- 3) Results of testing and discussion
- 4) Conclusions and future work



WiB Microlab

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- **Experimental**

- Set-up
- Slag comparison
- Sample preparation
- Thermogravimetric analysis



Results

- Chemically bounded water
- Calcium hydroxide
- Pore volume
- Compressive strength



Experimental: Plan

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- CEM I 52.5 R
 - Slag 1
 - Drinking water
- replacement rates:
3, 5, 10, 20, 30, 40, 60, 80, 95 mass %
w/b = 0.35 and 0.45
- CEM I 52.5 R
 - Slag 2
 - Drinking water
- replacement rates:
10, 20, 40, 60, 80, 100 mass %
w/b = 0.45
- CEM I **42.5 N**
 - **Woody ash**
 - Drinking water
- replacement rates:
10, 15, 20, 100 mass %
w/b = 0.5

Stop after 1 day, 7, 14, 28, 56, 365 days

Experimental: Slag comparison



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Components [mass %]	Slag 1	Slag 2	WA
CaO	42.44	41.22	41.3
SiO ₂	35.85	36.18	26.7
Al ₂ O ₃	11.38	12.13	4.72
MgO	6.00	7.23	7.5
S ²⁻	1.25	1.06	
TiO ₂	0.78	0.74	0.34
K ₂ O	0.37	0.55	7.86
Na ₂ O	0.22	0.39	1.95
Fe ₂ O ₃	0.41		2.04
Fe		0.22	
MnO	0.255		0.63
Mn		0.21	
Cl ⁻	0.014	0.04	0.054
SO ₃	0.17		0.68
Mn ₂ O ₃	0.283		

- CaO/SiO₂ (Slag 1) > CaO/SiO₂ (Slag 2)

- Al₂O₃ (Slag 2) > Al₂O₃ (Slag 1)

- MgO (Slag 2) > MgO (Slag 1)

+ Slag 2 is finer

	Slag 1	Slag 2
Blaine value [cm ² /g]	4000	4600

$$d_{50}(WA) = 146 \text{ um}$$

Experimental: Sample preparation



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1 Mixing

2 Storing

3 Hand-milling

4 Treating with acetone

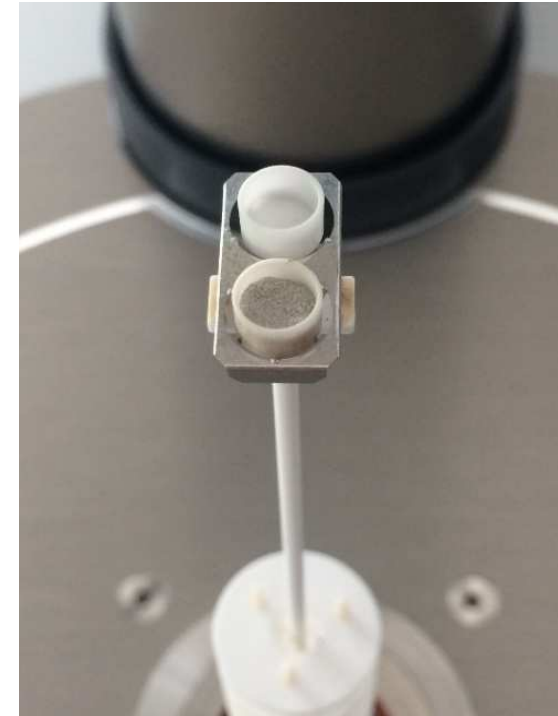


Experimental: Sample preparation



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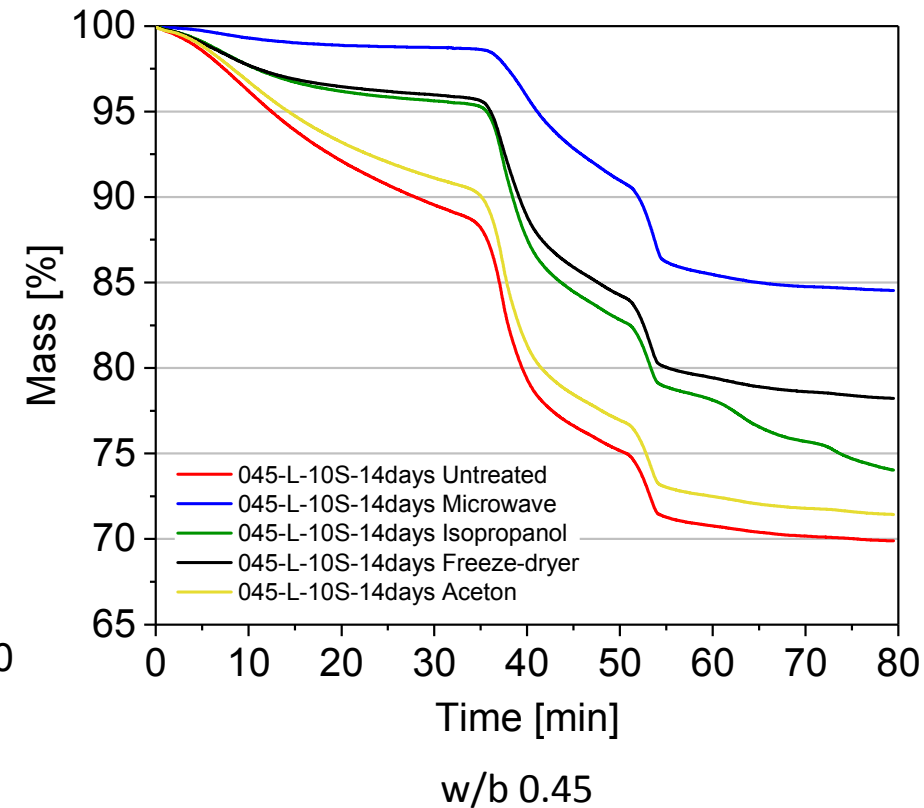
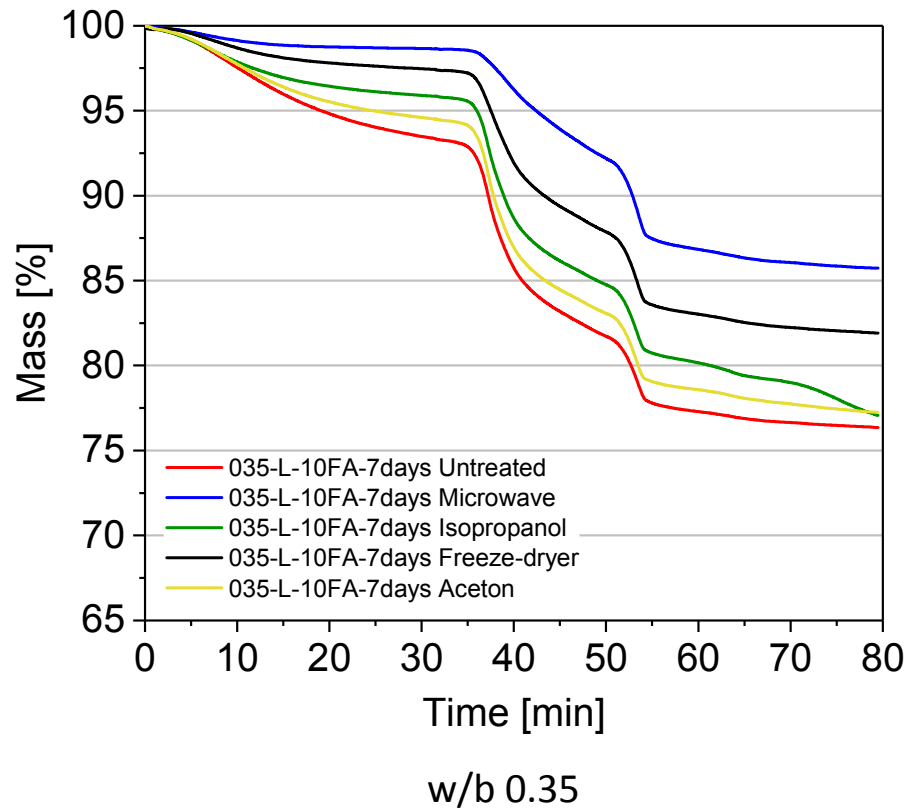
Before you start: When / How to stop hydration ?

Hydration Stoppage: Different Techniques



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Experimental: Thermogravimetric analysis

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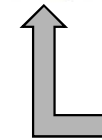
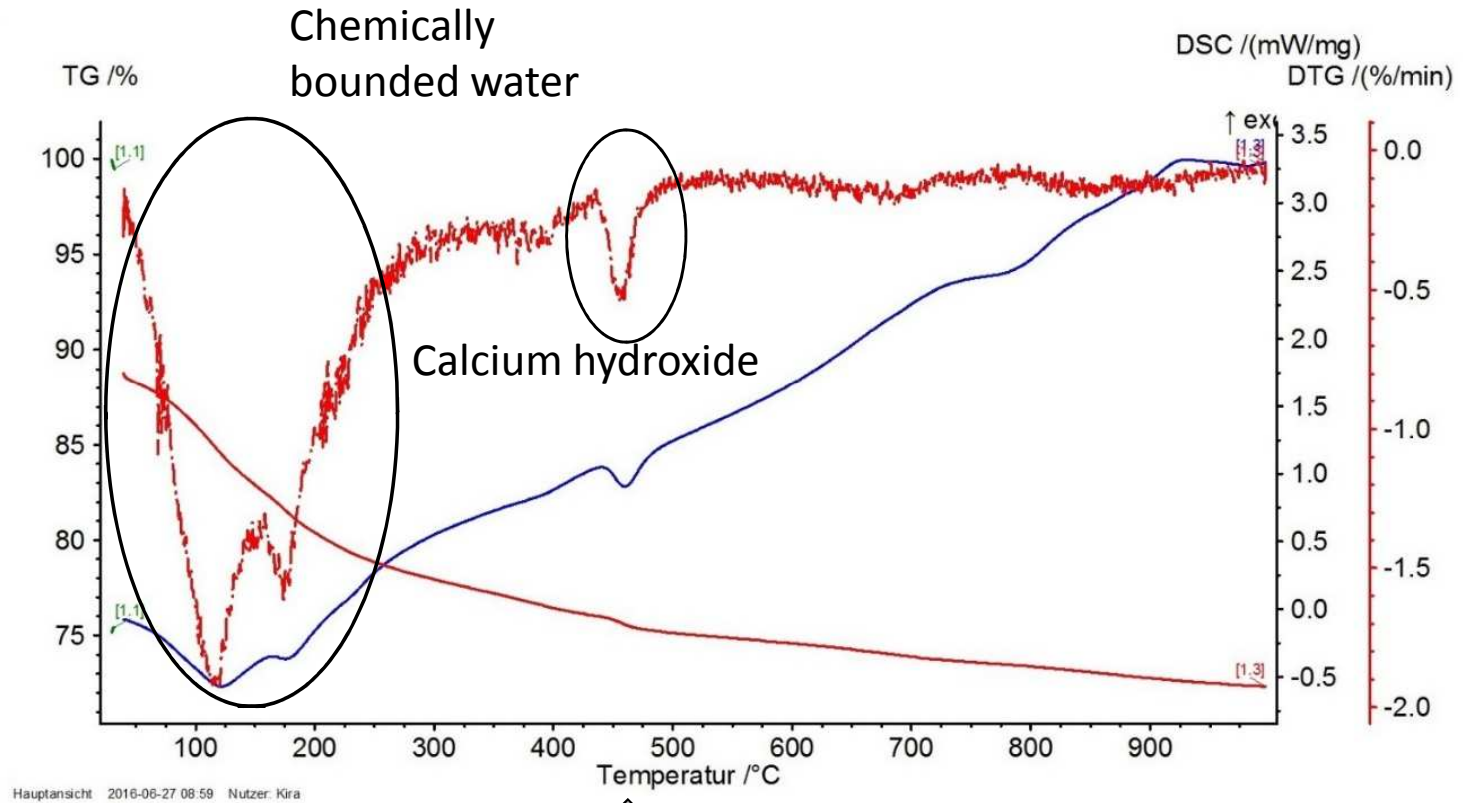
- Corundum crucible filled with 40 - 50 mg sample powder
- Nitrogen used as inert gas
- Temperature programme:
 - 30 minutes at 40 °C and
 - Heating to 1000 °C with a constant heating rate of 20 °C per minute



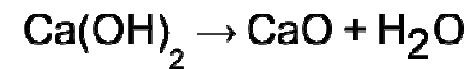
Results: Thermogravimetric analysis

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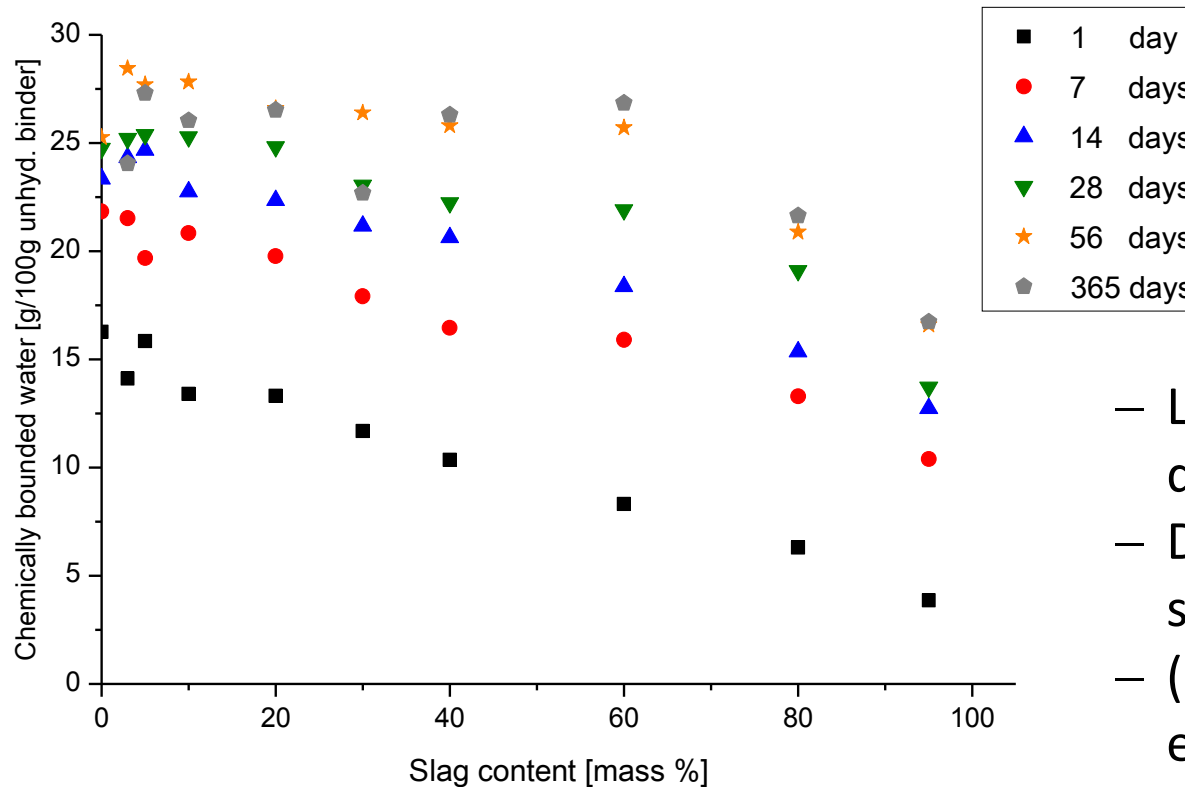
Ca(OH)_2 disintegrates between 400 – 500 °C



Results: Chemically bounded water (BW)



Chemically bounded water Slag 1 (w/b = 0.45)



- **BW removed from 40 - 600 °C**

- Largest increase from day 1 to 7
- Decrease with higher slag content
- (Slag > 20 mass % and early age)

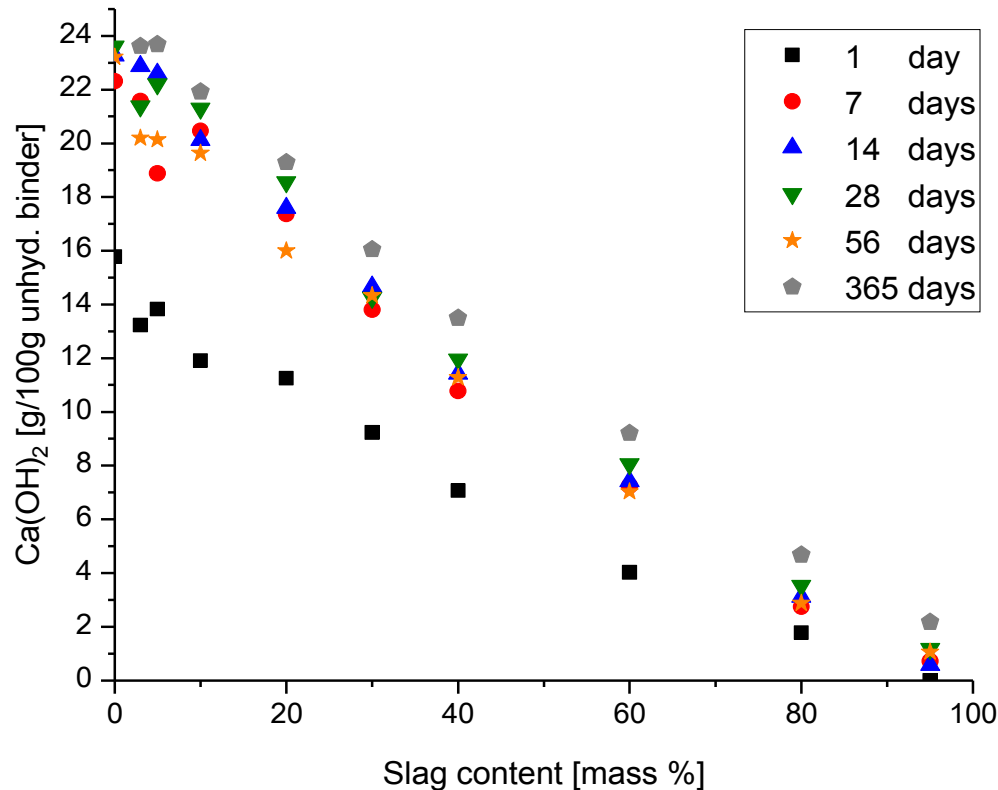
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Results: Calcium hydroxide (CH)



Ca(OH)₂ Slag 1 (w/b = 0.45)



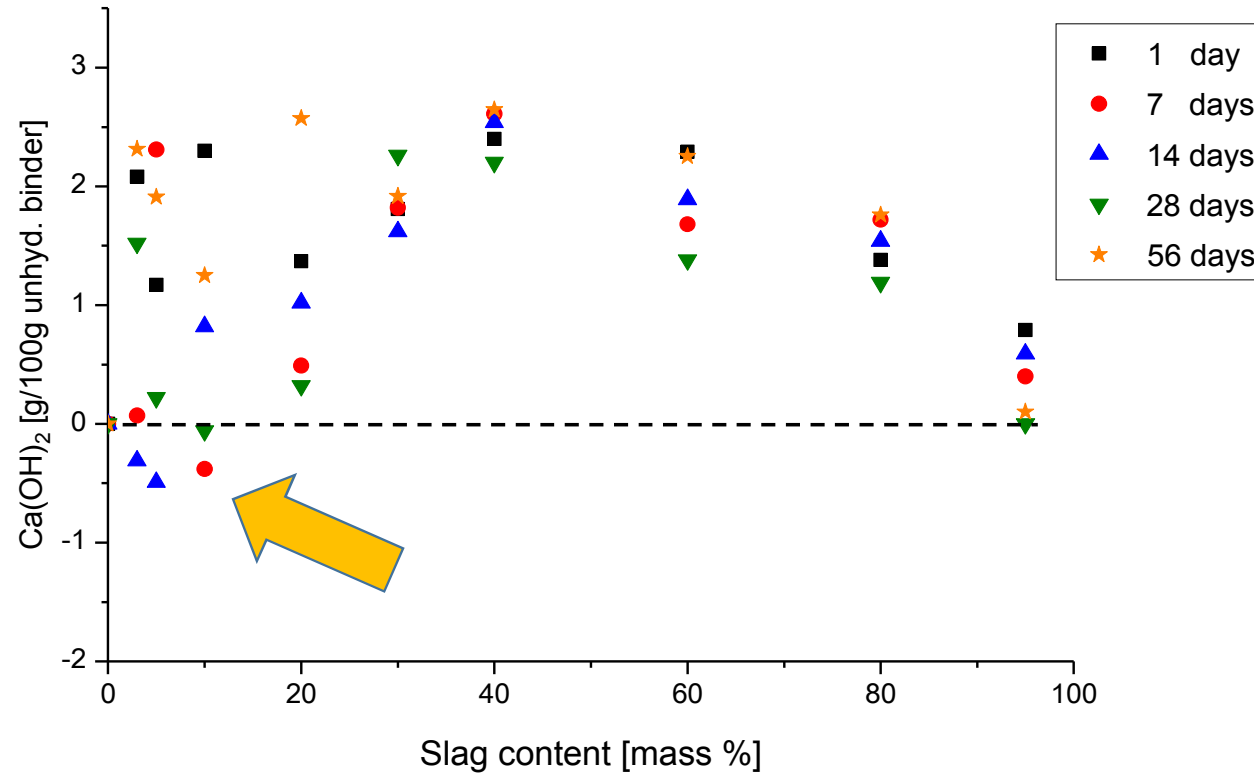
- CH decomposes from **360 - 550 °C** (tangential method was used)
- CH content decreases with increasing slag content
- CH content increases with time
- w/b = 0.35 lower CH content when slag content ≤ 20 %

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Results: Consumption of calcium hydroxide



Consumption of Ca(OH)₂ Slag 1 (w/b = 0.45)



Consumption = Ca(OH)₂ produced by cement (reference) - Ca(OH)₂ measured

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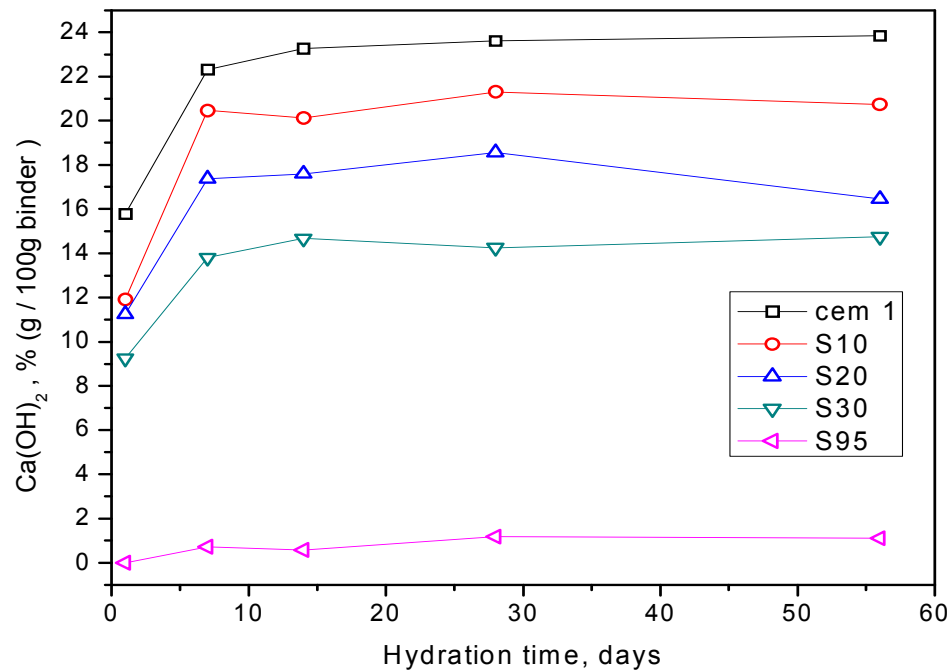
Results: Calcium hydroxide slag (S) versus Woody ash (WA)



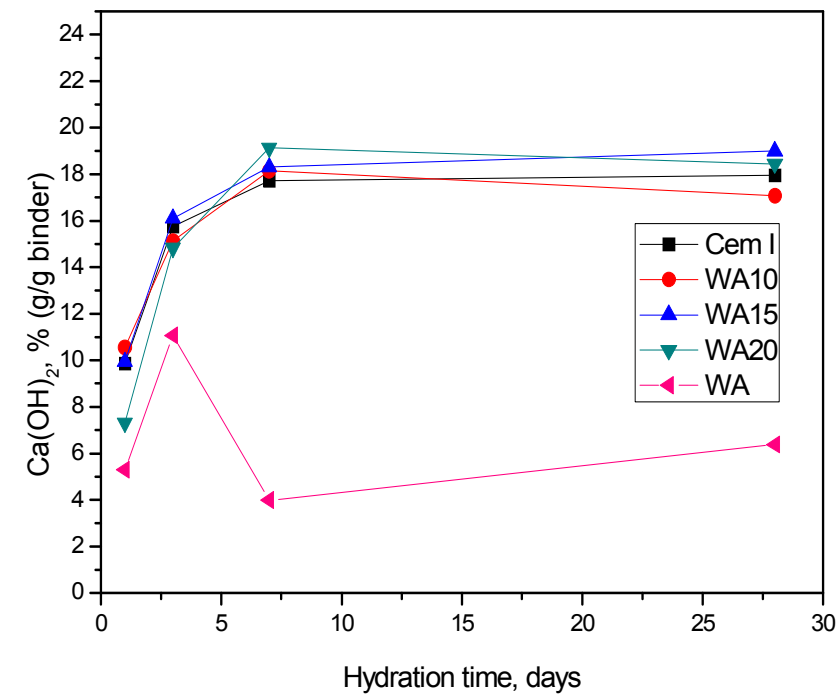
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Ca(OH)₂: mas.% per powder (cem I + slag 1)



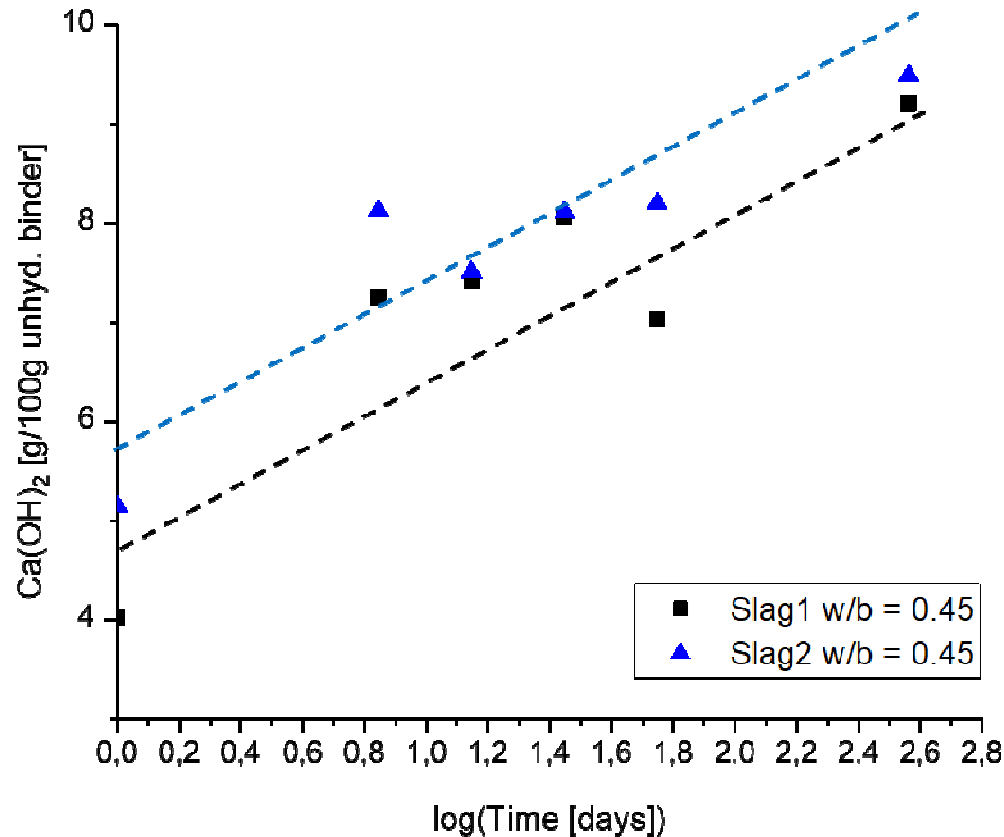
Ca(OH)₂: mas.% per powder (cem I + WA)



Results: Calcium hydroxide (CH)



Ca(OH)₂ (60 % slag)



- Logarithmic scale
- CH content increasing linearly with log of time
- Slag 2 slightly higher CH content

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Results: Calcium hydroxide regression model



Regression model:

$$CH = \beta_0 + \beta_1 \cdot \text{Slag} + \beta_2 \cdot \log(t)$$

CH = Ca(OH)₂ [g / 100g unhyd. binder]

Slag = slag content of initial mixture [mass %]

t = hydration time [days]

Slag	w/b	β_0	β_1	β_2	R ²
Slag1	0.35	15.56	-0.19	1.66	0.9556
Slag1	0.45	16.73	-0.22	1.83	0.9462
Slag2	0.45	18.14	-0.22	1.45	0.9604

Results up to 28 days included.

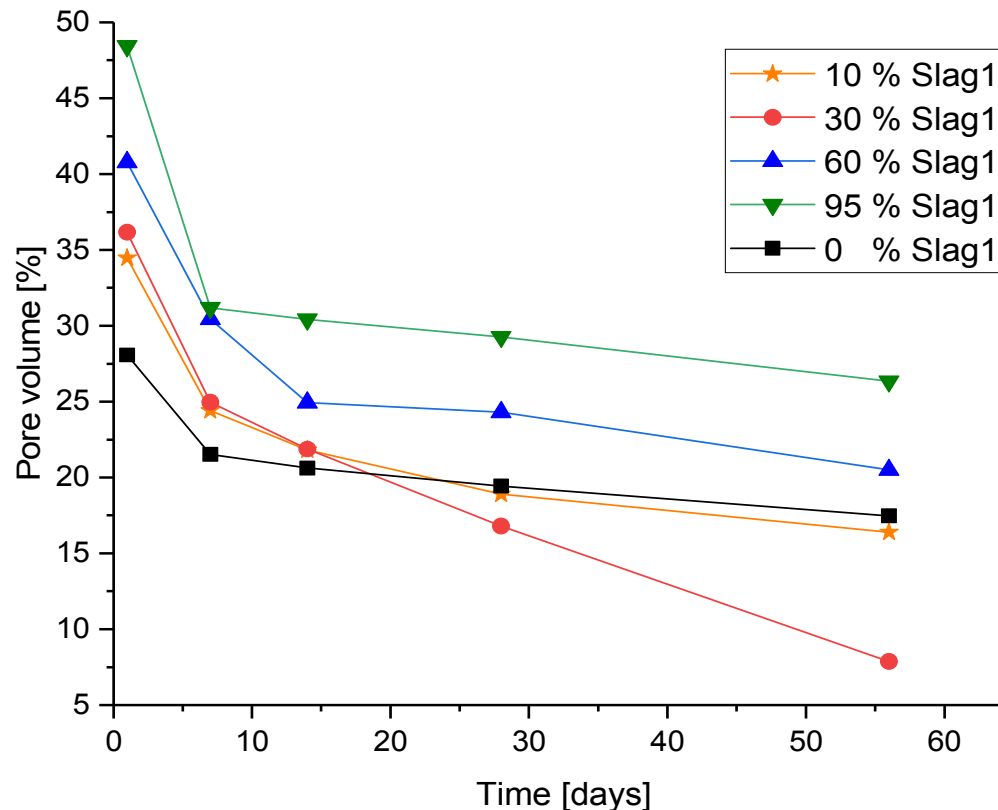
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Additional results: Pore volume



Pore volume Slag 1 (w/b = 0.35)

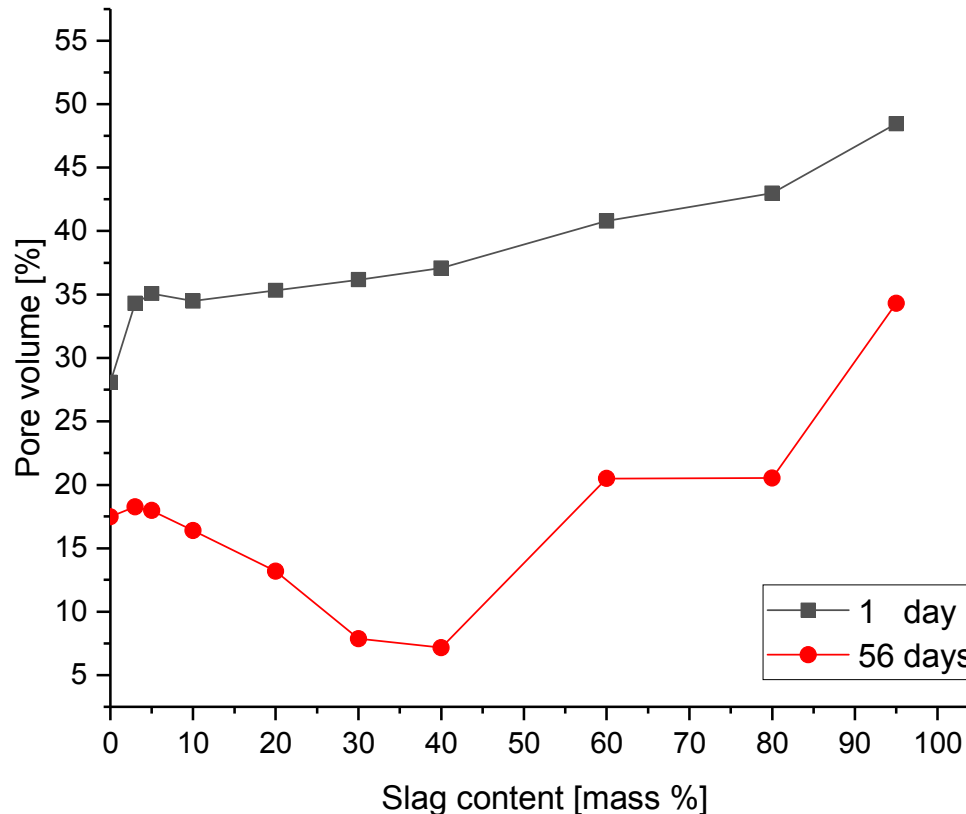


- Pore volume measured with pycnometer
- Decreasing pore volume with time
- Porosity increases for slag content > 30 %

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Additional results: Pore volume

Pore volume Slag 1 (w/b = 0.35)



- After 1 day higher porosity with increasing exchange rate
- After 56 days the minimum calculated porosity is with 40 % exchange rate

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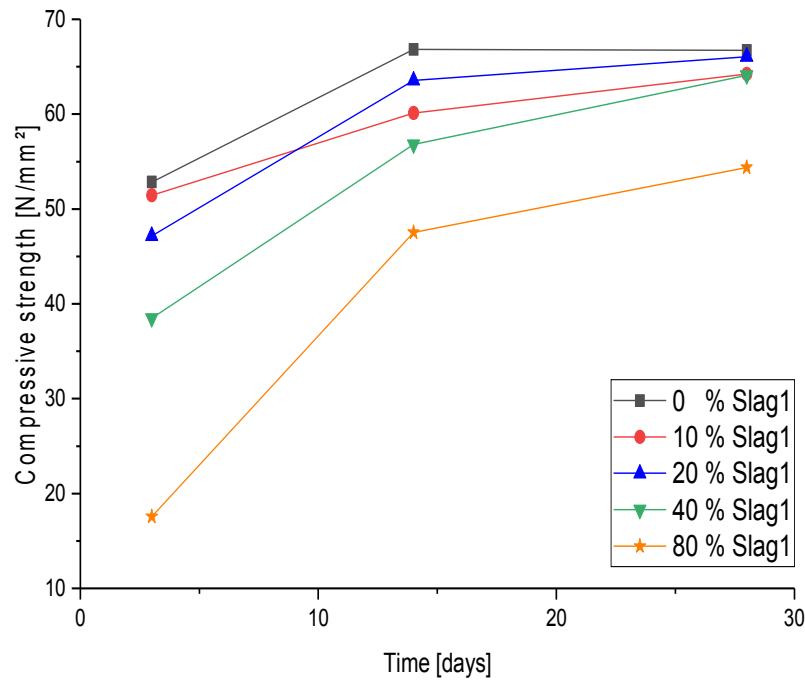
Additional results: Compressive strength



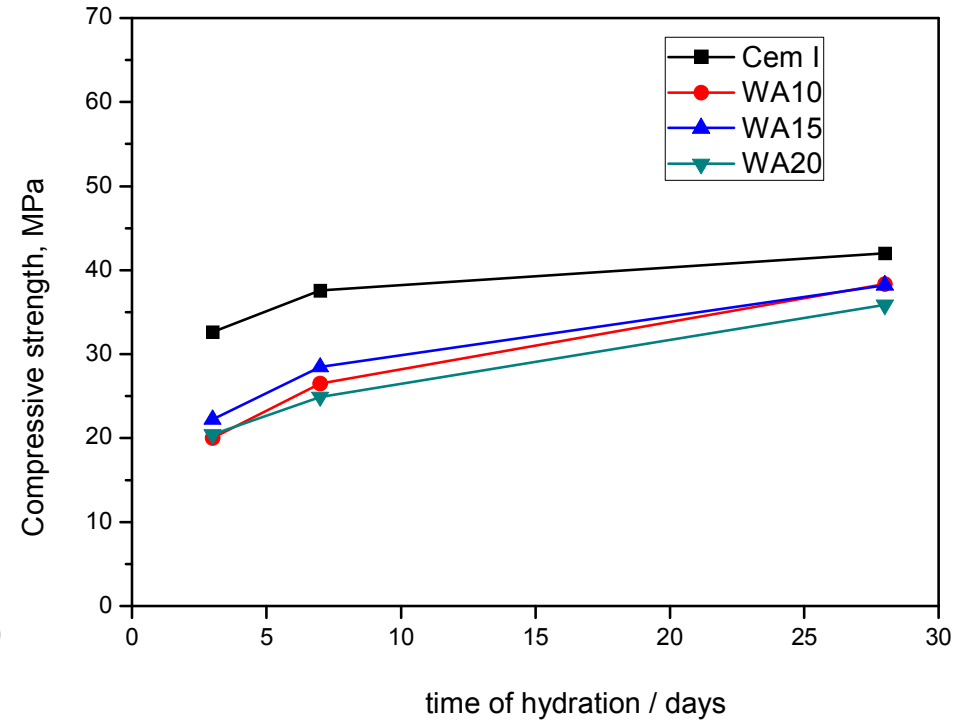
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Compressive strength Slag 1 (w/b = 0.35)



Woody ash (w/b = 0.5)



Conclusions



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- Ca content of slag and WA are nearly the same
- CH consumption for WA has almost negligible up to 20%
- Reactivity of WA differs from slag
- Pore structure depending on replacement ratio
- For WA replacement rate up to 20% has constant impact on strength

Greetings from Darmstadt!

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THANK YOU FOR YOUR ATTENTION

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Transformation of Wood Biomass Ash into Resilient Construction Composites